

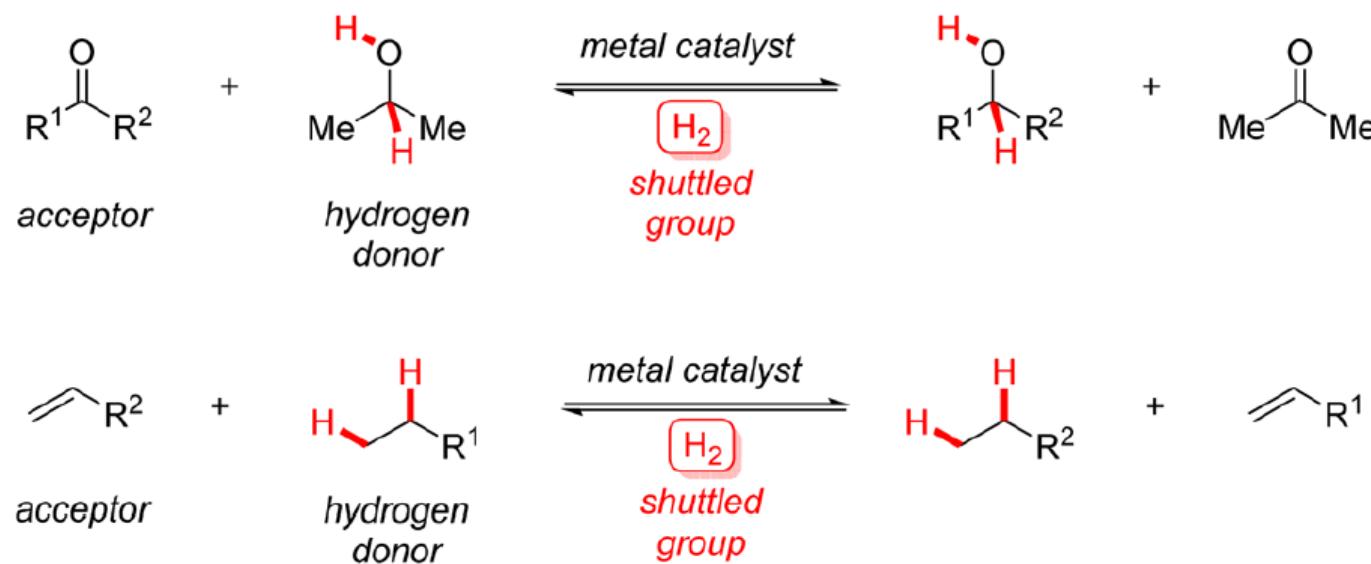
Catalytic Transfer Functionalization through Shuttle Catalysis

Reporter: He Zhiqi
Supervisor: Prof. Yong Huang
2017.06.26

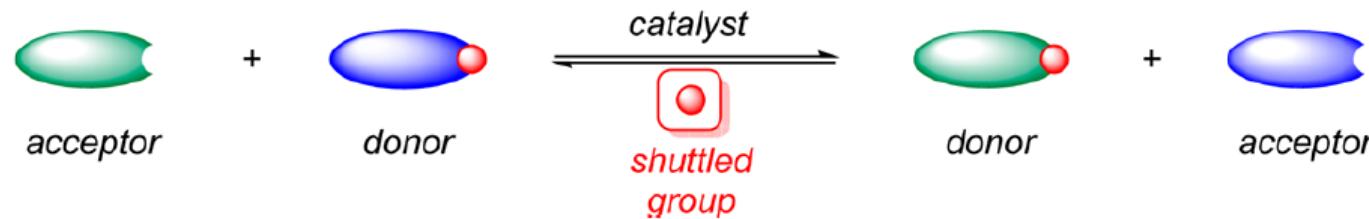
Morandi. B., *et al.* *ACS Catal.* **2016**, *6*, 7528
Morandi. B., *et al.* *Chem. Eur. J.* **2017**, DIO: 10.1002/chem.201605325

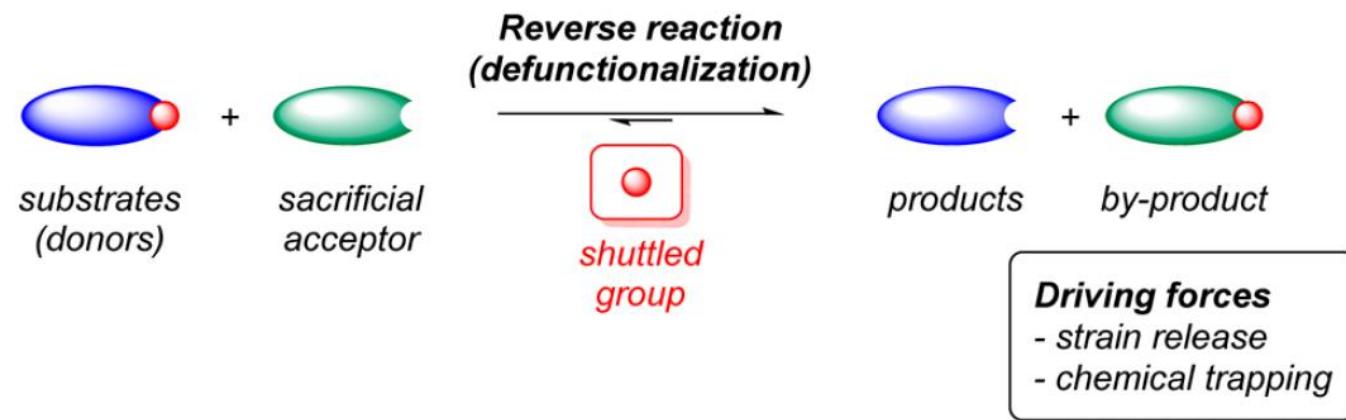
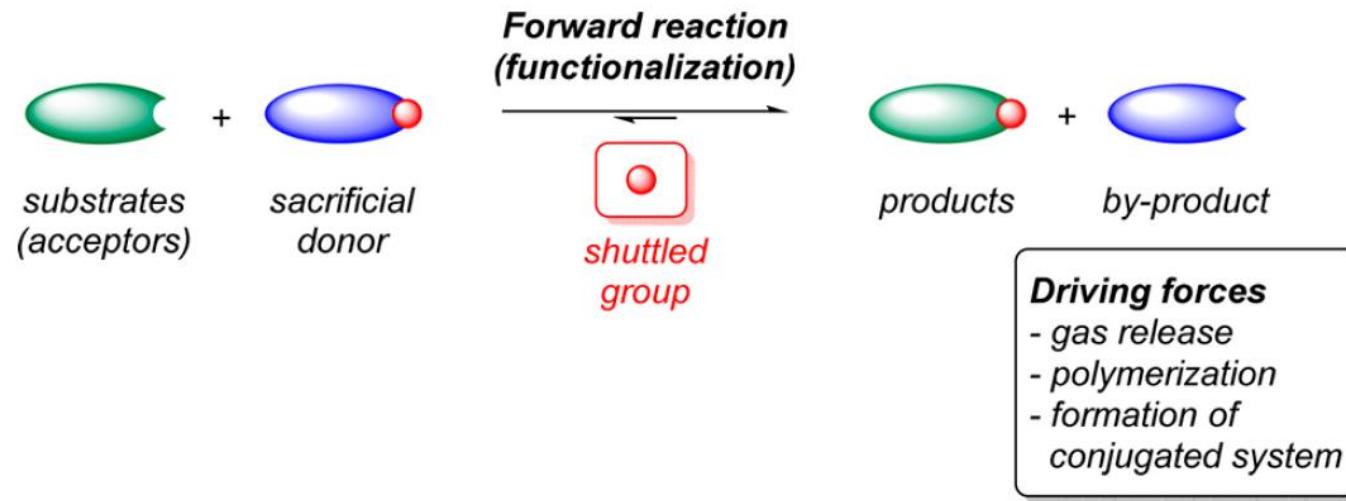
1. Introduction

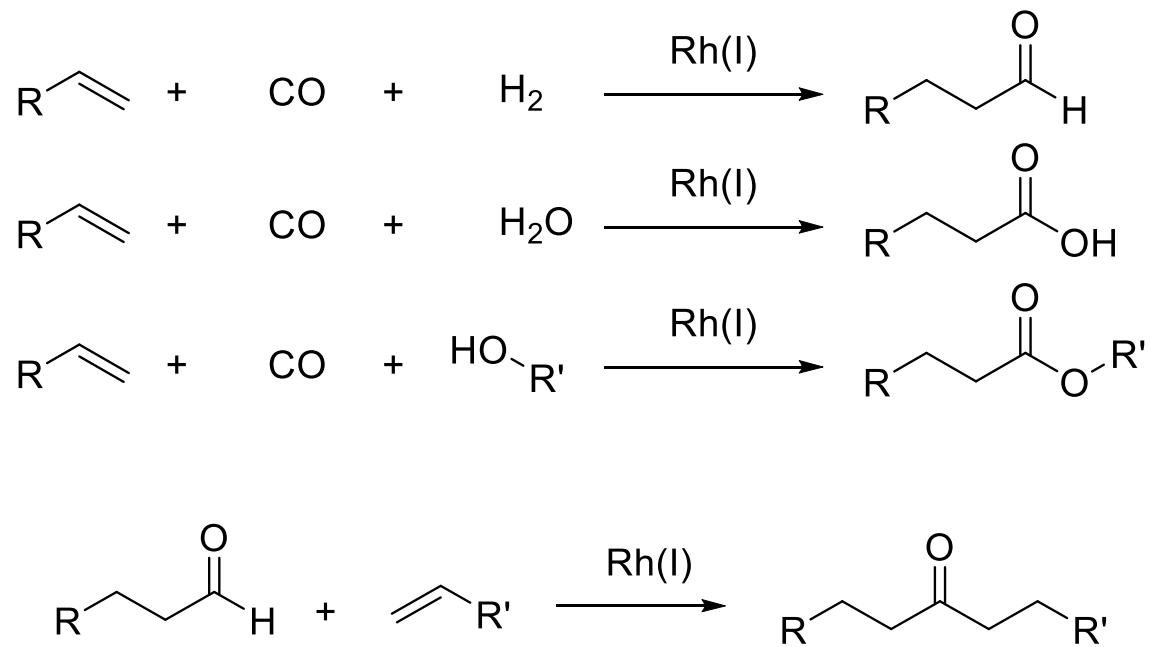
a) Transfer Hydrogenation (H_2 -shuttle)



b) Generalization of the Shuttle Catalysis Concept

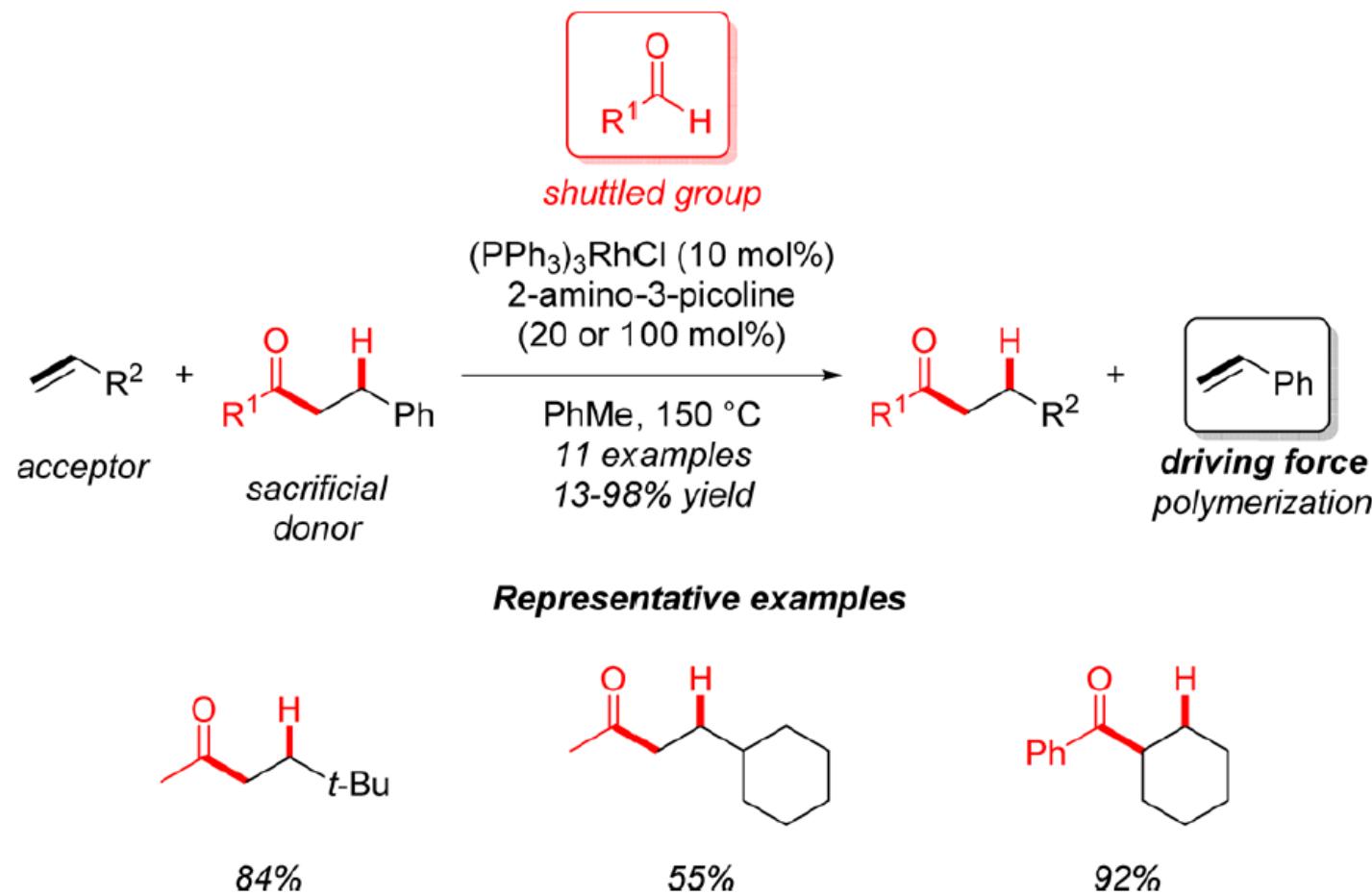






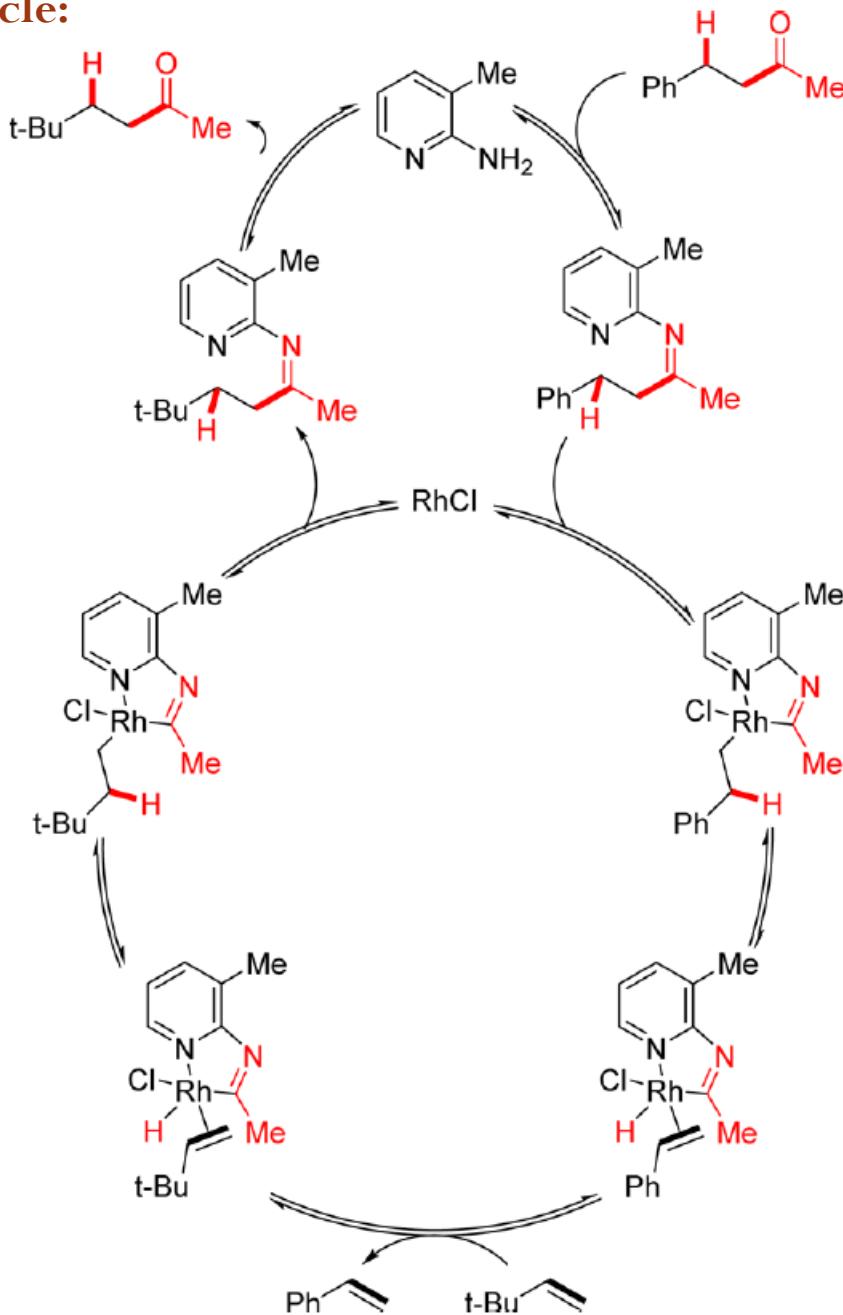
Willis, M. C. *Chem. Rev.* **2010**, *110*, 725

2. Seminal examples of shuttle catalysis beyond hydrogen transfer:

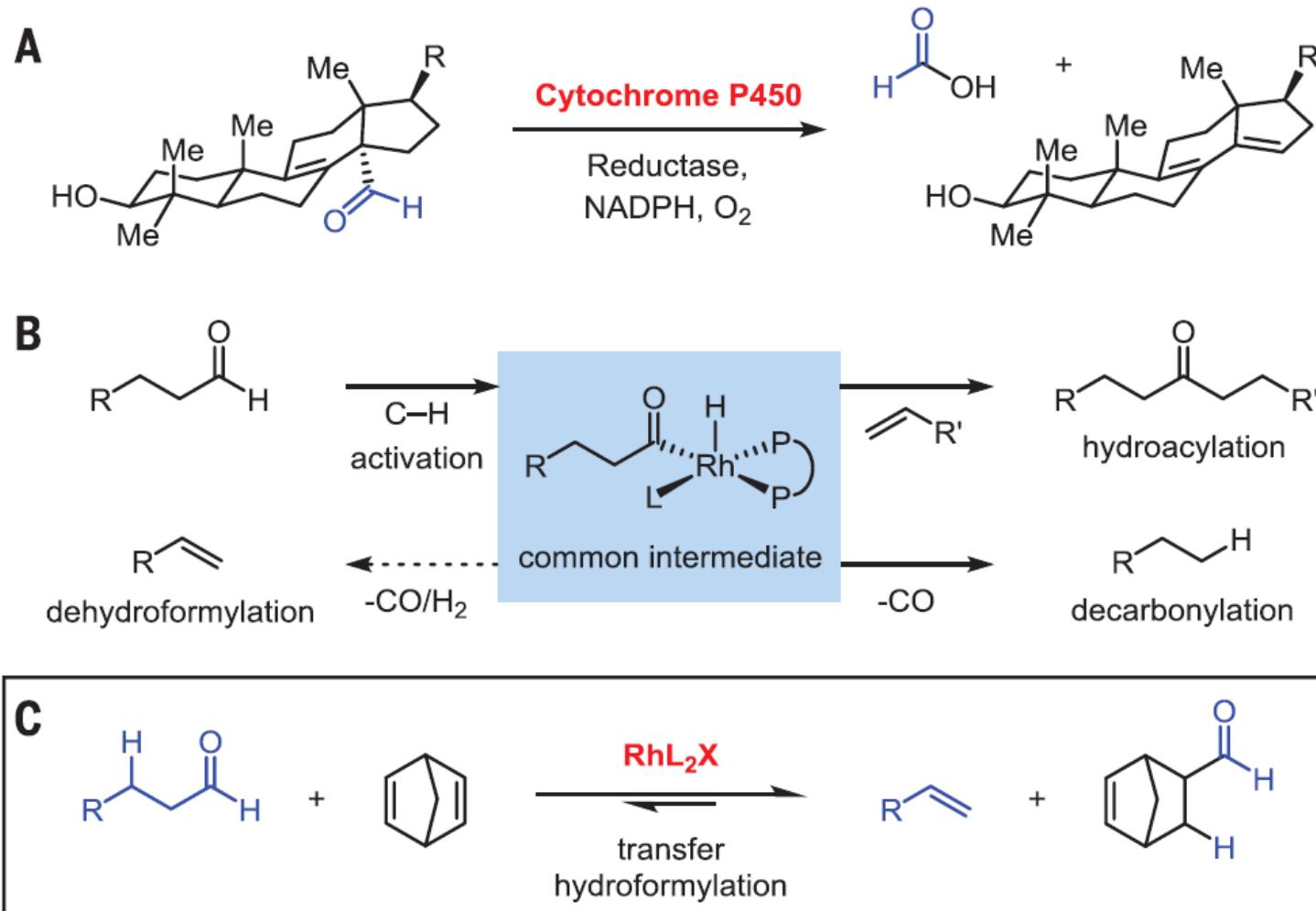


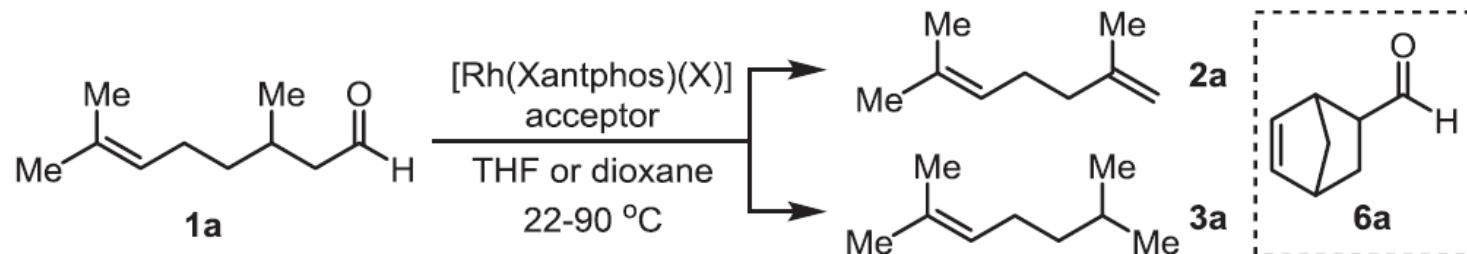
Jun, C.-H., et al. *J. Am. Chem. Soc.* **1999**, *121*, 880

Proposed catalytic cycle:



3. Rhodium catalyzed retro-hydroformylation:



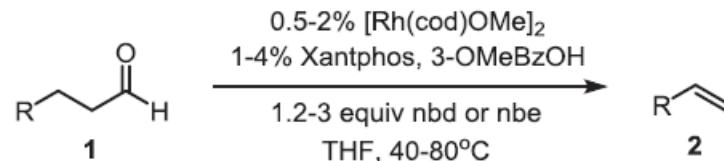
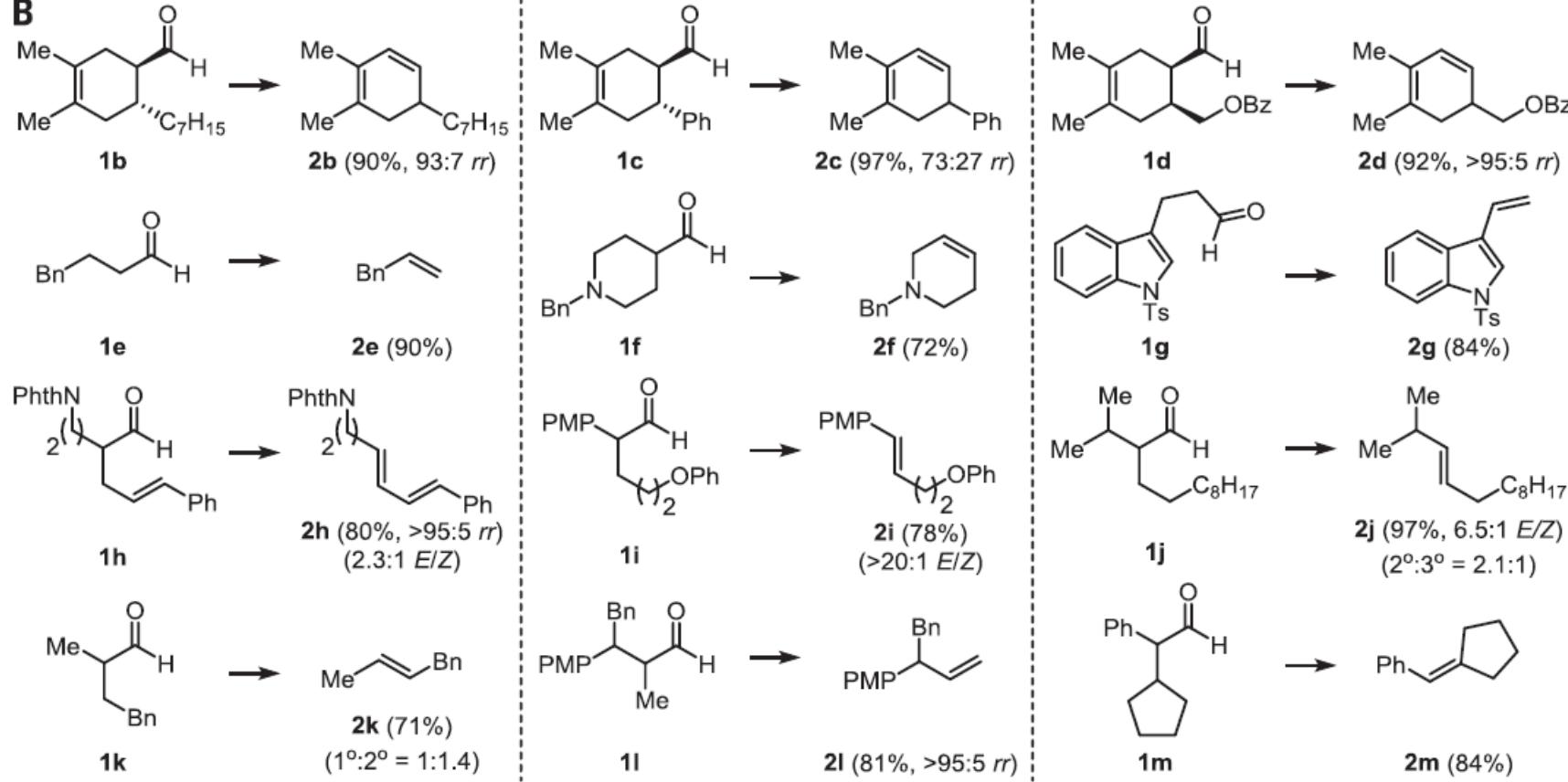


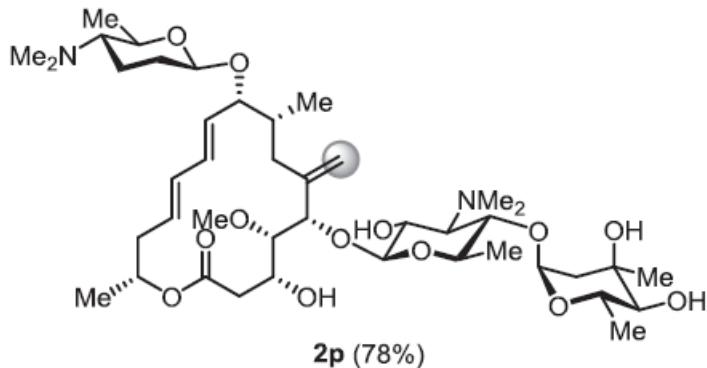
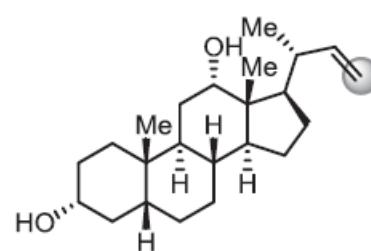
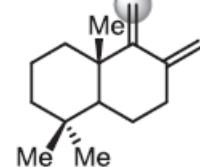
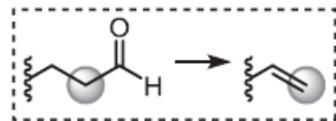
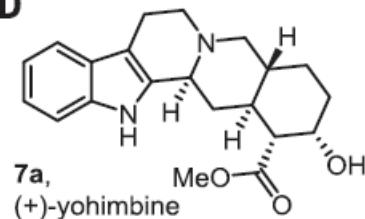
counterions (yield, **2a**:**3a**): 10% catalyst, 1.2 equiv nbd, dioxane, 90 °C, 1h

4a 0%	4b 0%	4c 6%, 85:15	4d 27%, 98:2	4e 27%, 98:2	4f 99%, >99:1

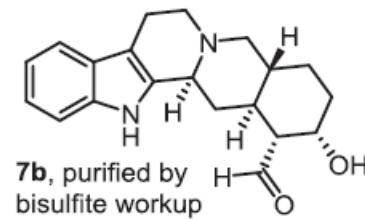
acceptors (yield, **2a**:**3a**): X = 3-OMeBzO⁻, 1.2-3 equiv acceptor, THF, 22-80 °C

5a, nbd 88%, >99:1 0.3% [Rh], 80 °C	5a, nbd 99%, >99:1 1% [Rh], 60 °C	5b, nbe 97%, 99:1 2% [Rh], 40 °C	5c, bnbd 99%, 99:1 2% [Rh], 22 °C

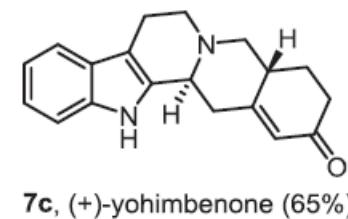
A**B**

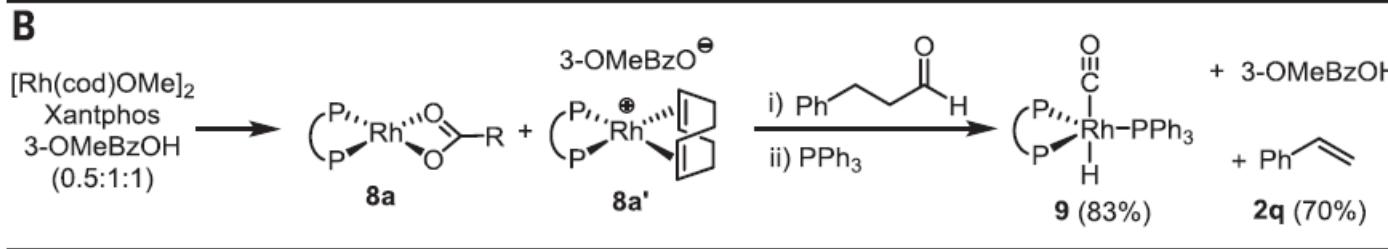
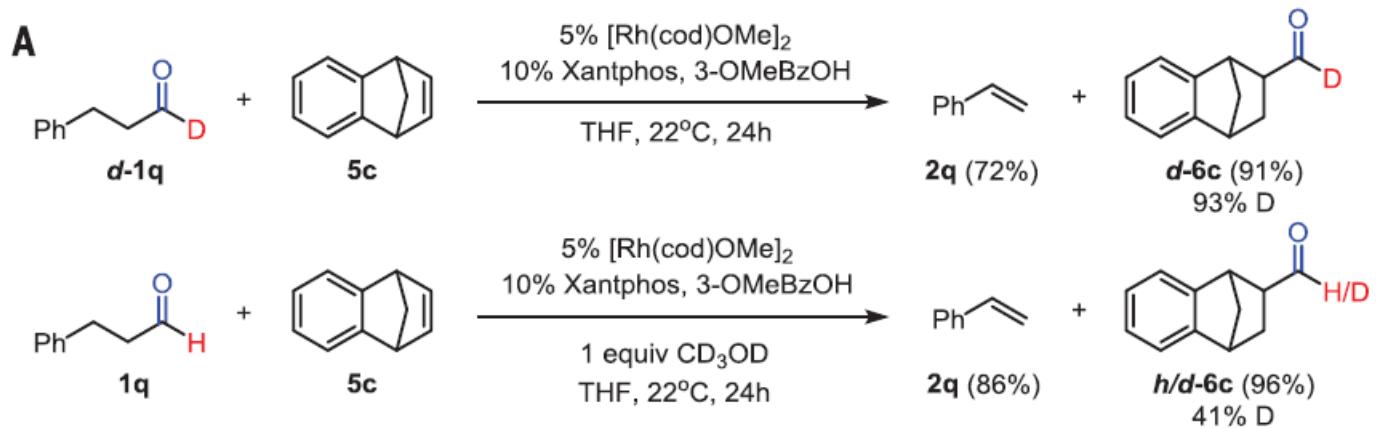
C**D**

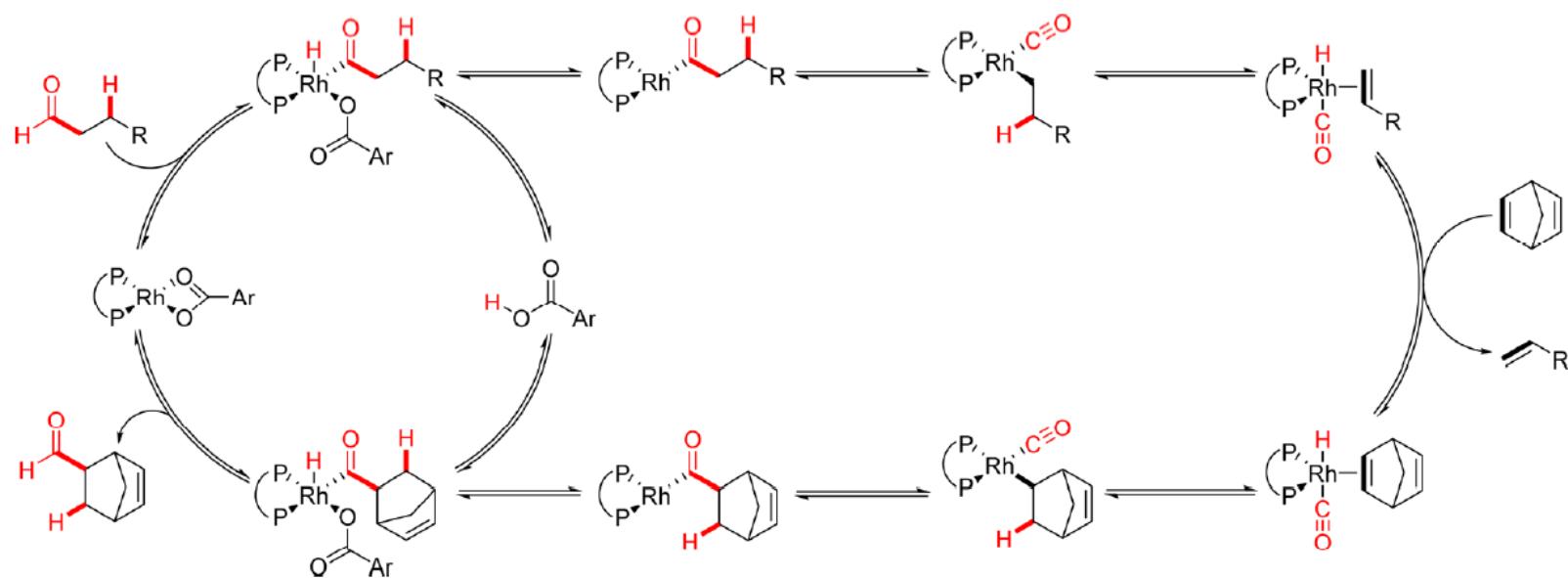
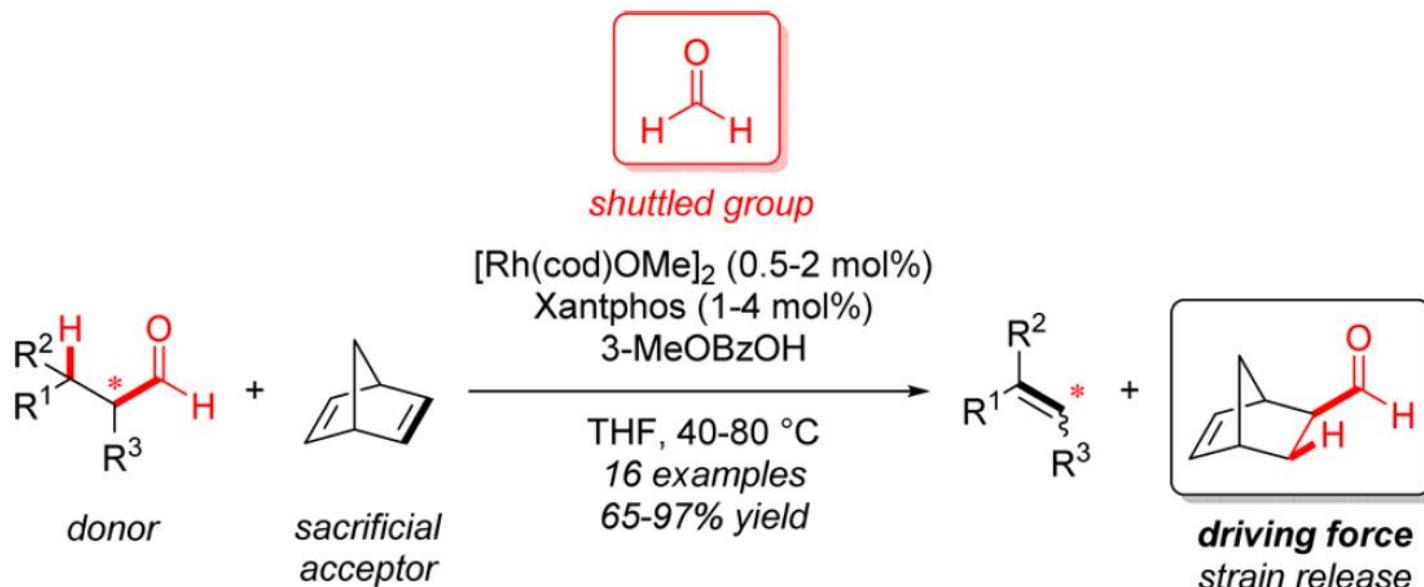
i) LiAlH₄, 94%
ii) SO₃-pyridine,
DMSO, NEt₃,
93%



2% [Rh(cod)OMe]₂
4% Xantphos
4% 3-OMeBzOH
6 equiv nbe
70°C, THF, 24h

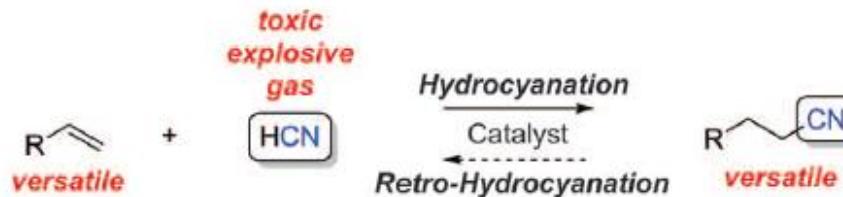






4. Nickel-catalyzed transfer hydrocyanation reaction:

A Traditional Approach to Interconvert Alkenes and Nitriles



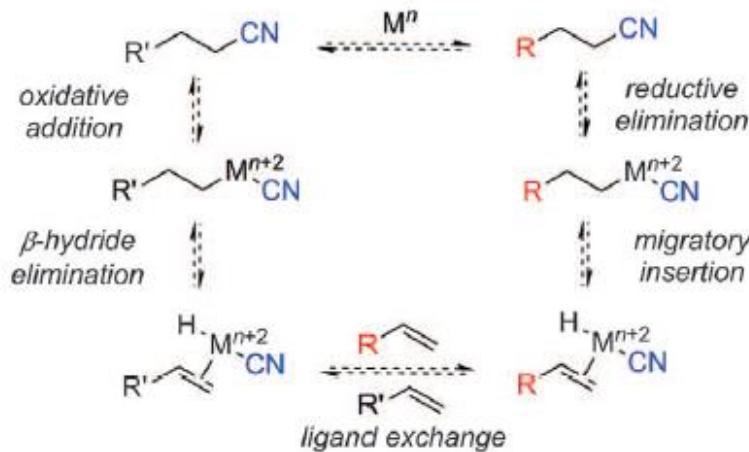
Hydrocyanation: underexploited in routine chemical synthesis

Retro-Hydrocyanation: challenging, thermodynamically uphill

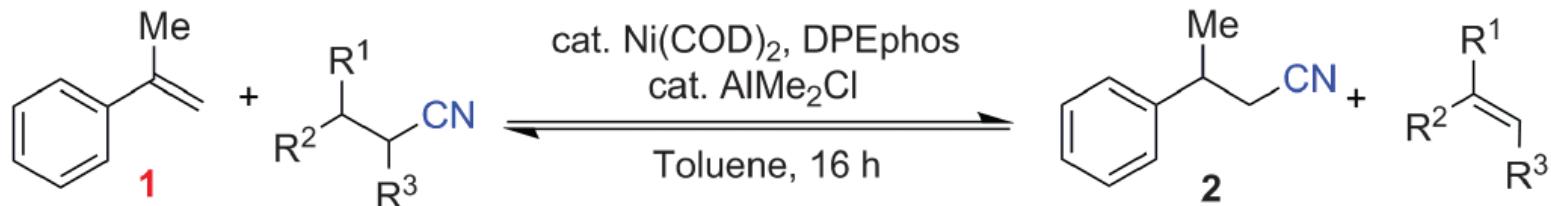
B Our Design: **HCN-free** reversible transfer hydrocyanation



Proposed Mechanism



A Thermodynamic Challenge: How can we drive the equilibrium to obtain **1** or **2** selectively ?



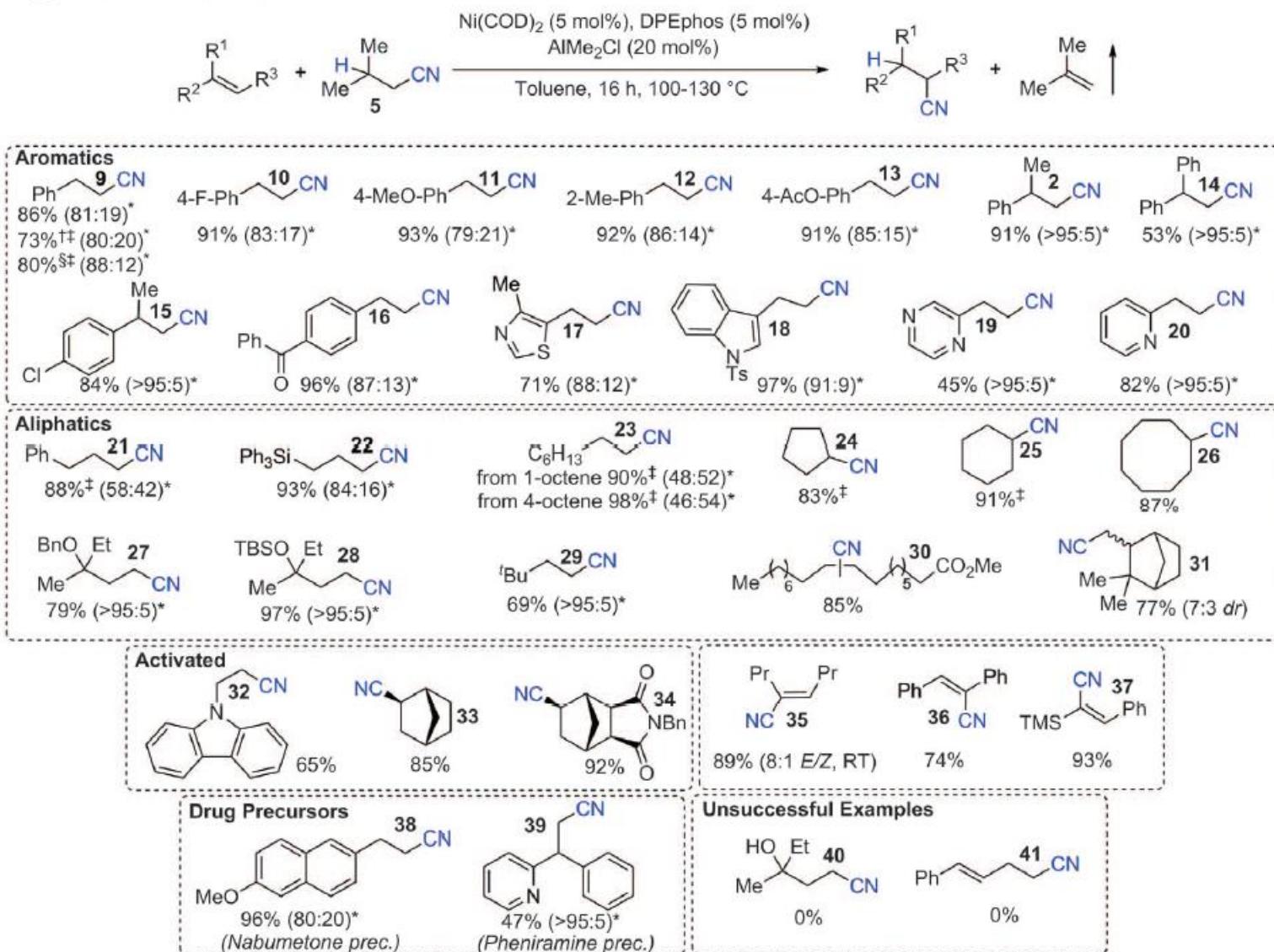
B Hydrocyanation: Formation of gaseous disubstituted alkene best driving force

R^1 R^2 - <chem>CC(R3)C(CN)C</chem> R ³ (5 equiv.)	3	4	5	6
yield 1 \rightarrow 2 (100 °C)	3%	26%	69%	60%
<i>open system</i>	—	41%	86%	67%

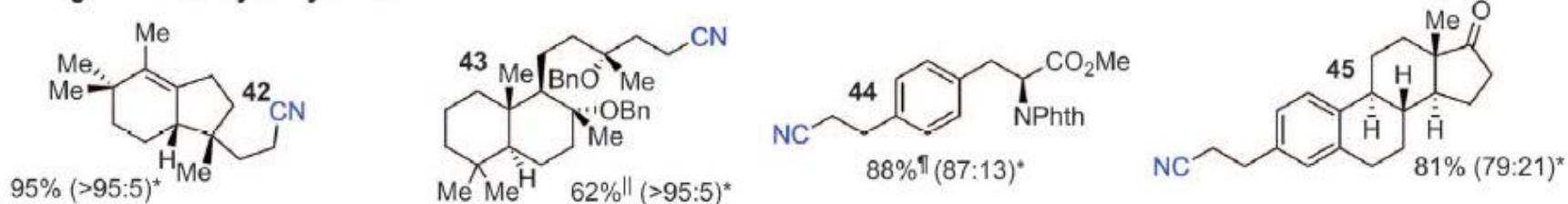
C Retro-Hydrocyanation: Strained alkenes best driving force

R^1 R^2 - <chem>CC(R3)=C</chem> R ³ (1 equiv.)	no acceptor	7	8
yield 2 \rightarrow 1 (28 °C)	<5%	46%	99%

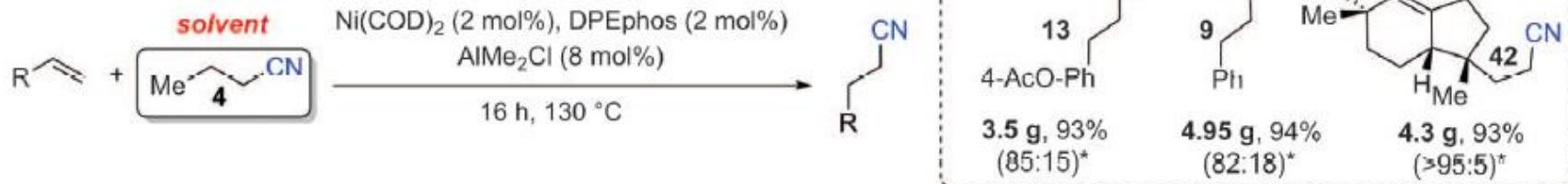
A Scope of the Hydrocyanation

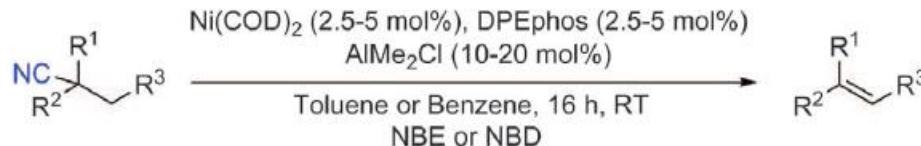


B Late-stage Transfer Hydrocyanation



C Scale-up Using an Inexpensive Reagent

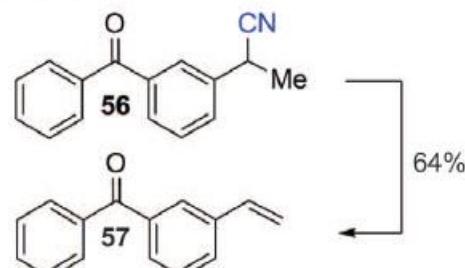
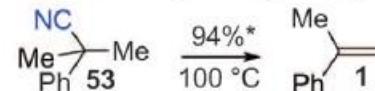
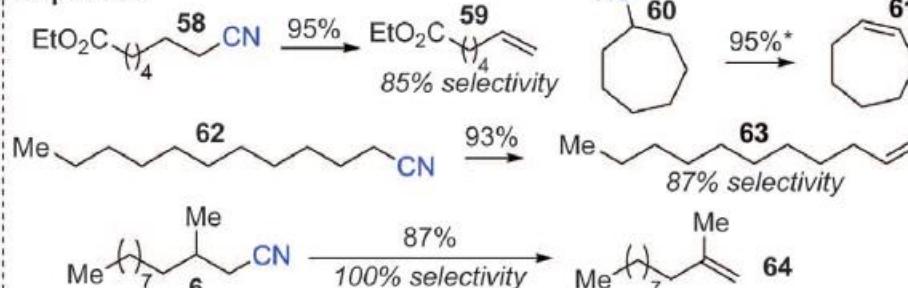
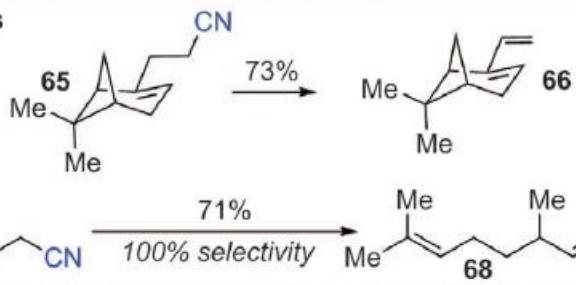
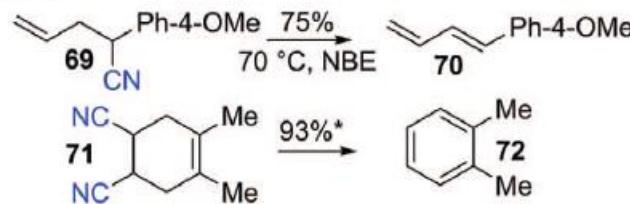


A**Styrene Synthesis**

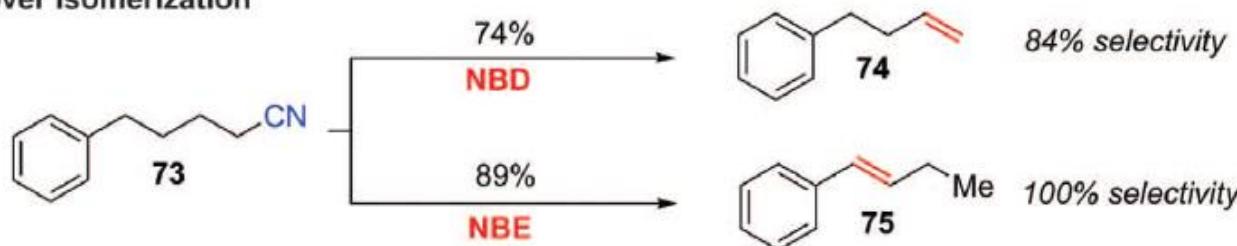
$\text{Ar}^1 = \text{Ar}^2 = \text{Ph}$ (**46->47**); 99%
 $\text{Ar}^1 = 3\text{-CF}_3\text{-Ph}; \text{Ar}^2 = \text{Ph}$ (**48->49**); 97%
 $\text{Ar}^1 = 3\text{-Cl-Ph}; \text{Ar}^2 = \text{Ph}$ (**50->51**); 90%



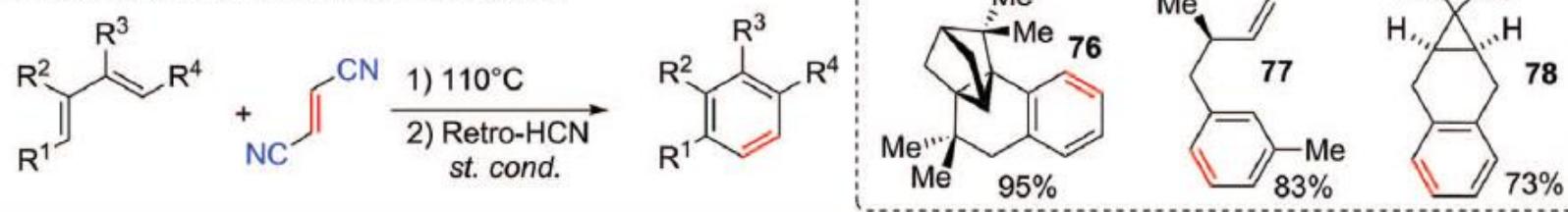
$\text{R} = \text{Me}; \text{Ar} = \text{Ph}$ (**2->1**); 99%*
 $\text{R} = \text{Ph}; \text{Ar} = \text{Ph}$ (**14->52**); 96% (70 °C)

**Aliphatics****Terpene Derivatives****Polyene Products**

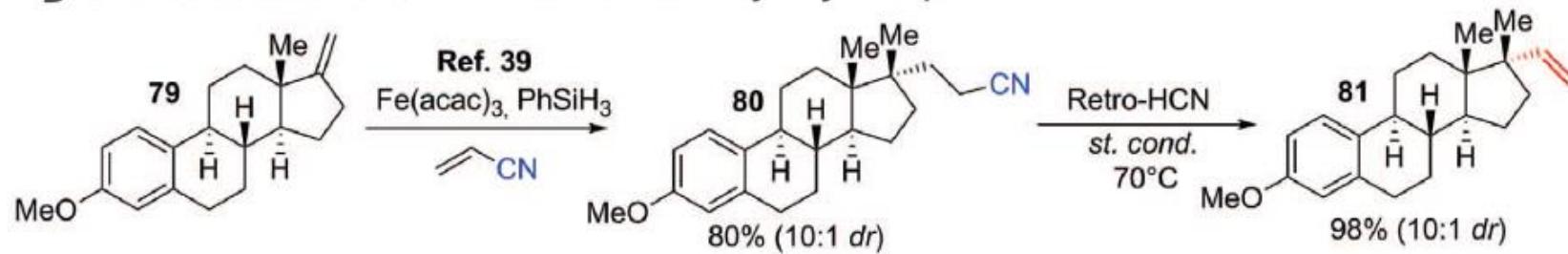
B Control over Isomerization

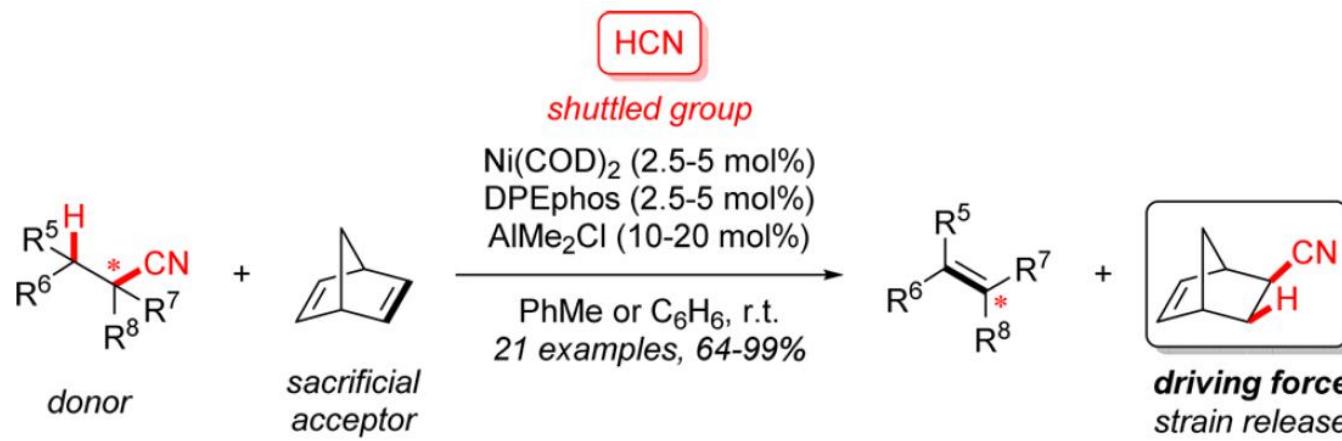
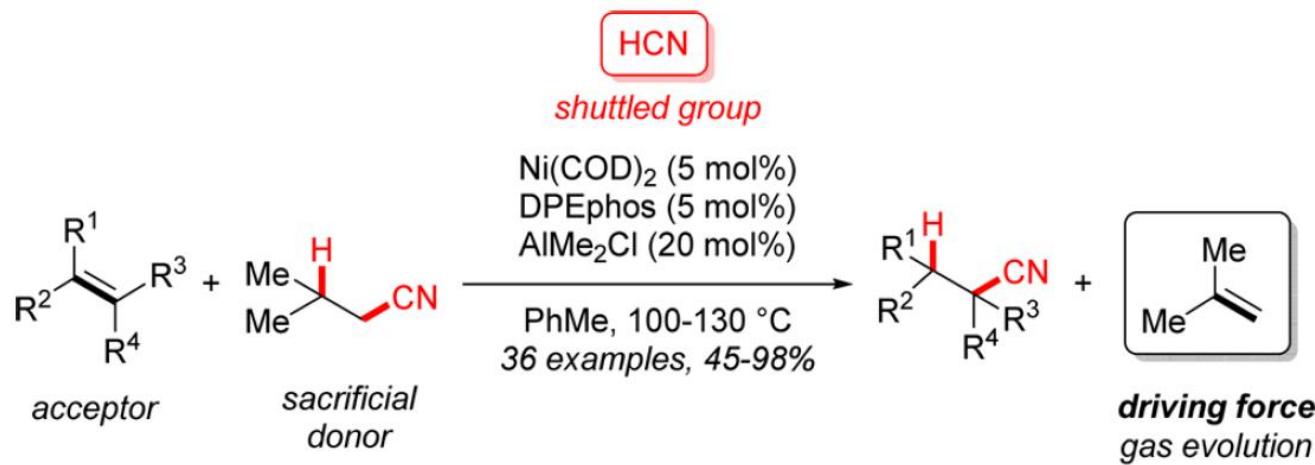


C Construction of Aromatics from Dienes



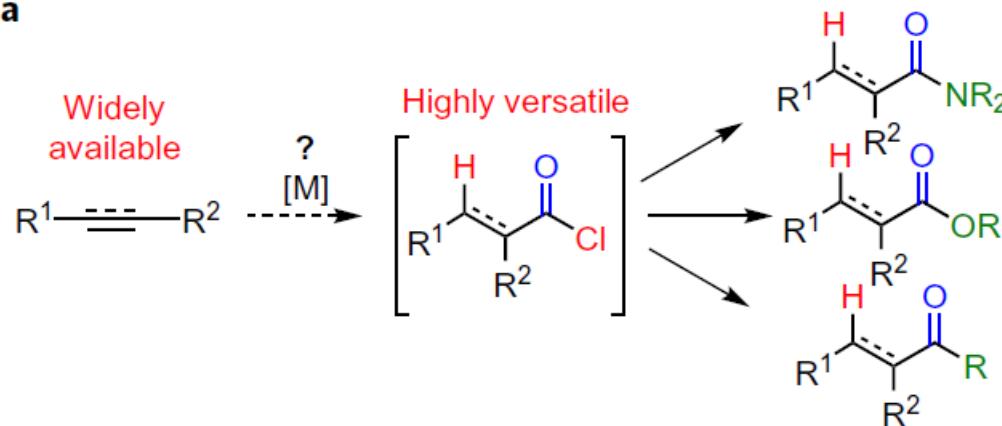
D Stereoselective Installation of a Chiral Quaternary Vinyl Group



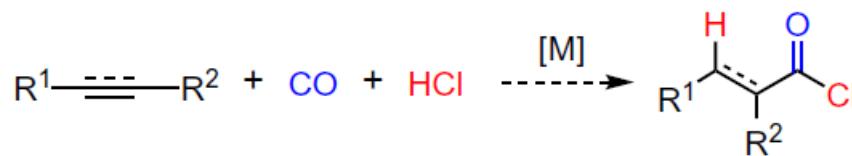


5. CO- and HCl-free synthesis of acid chlorides via Palladium catalyzed

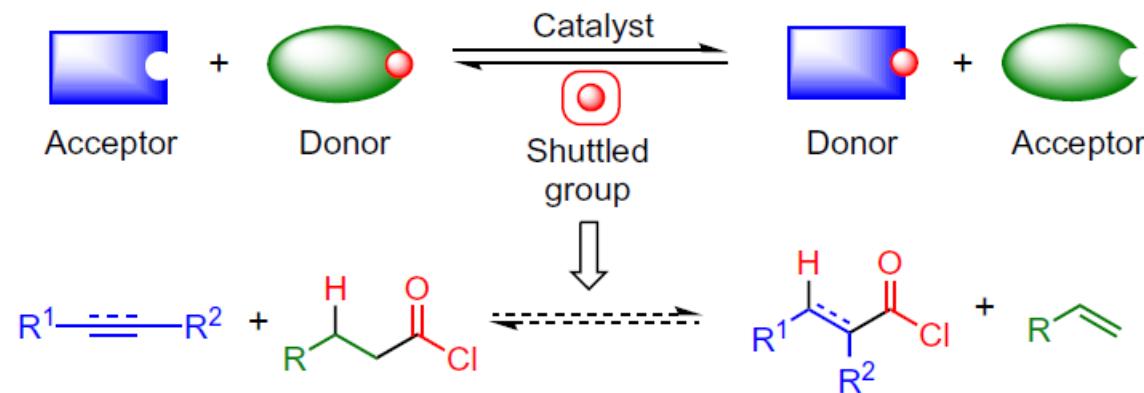
a



b



c



d

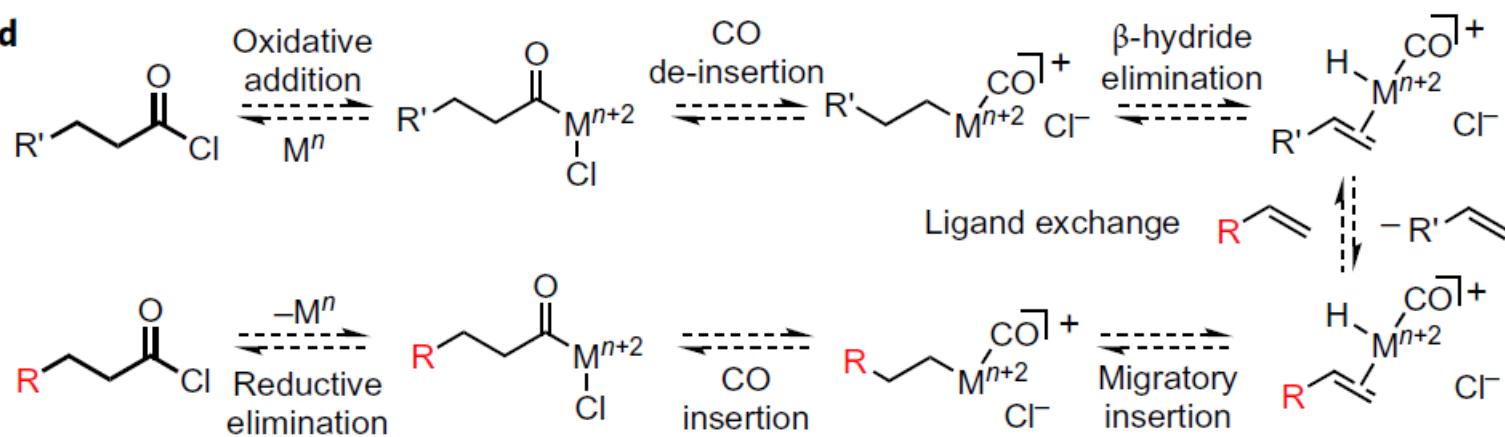


Table 1 | Scope of the transfer hydrochlorocarbonylation.

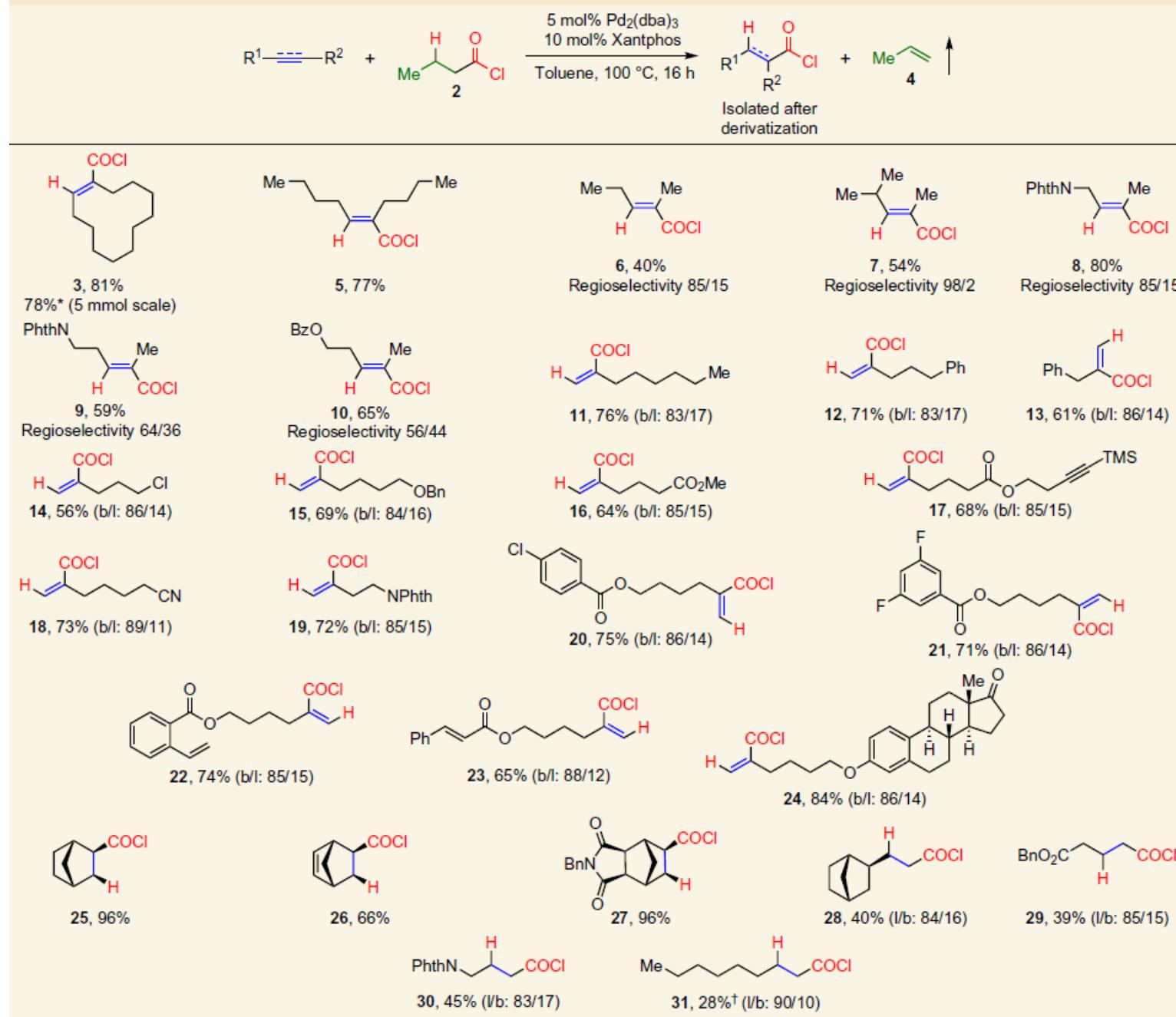
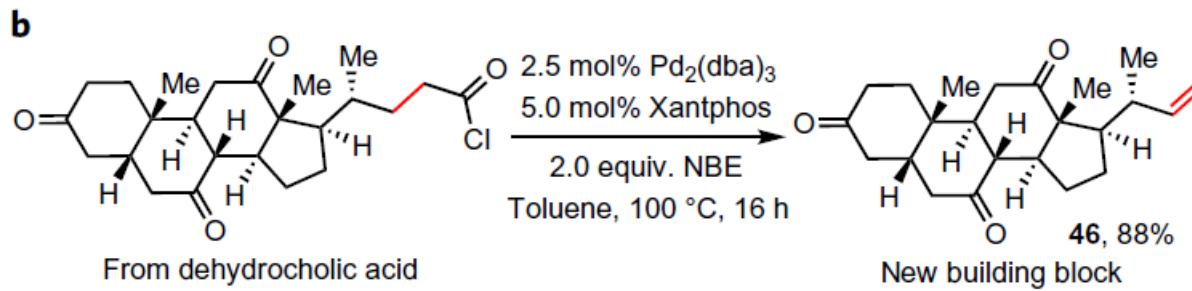
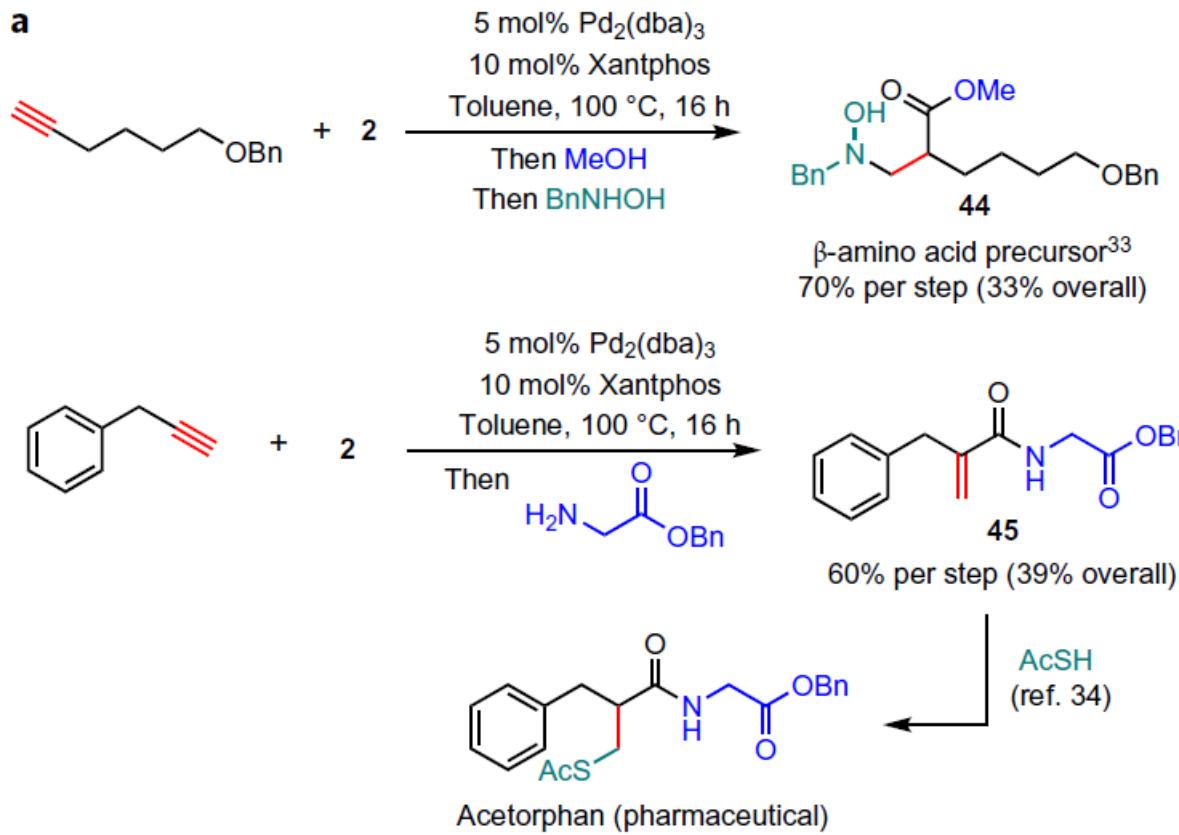
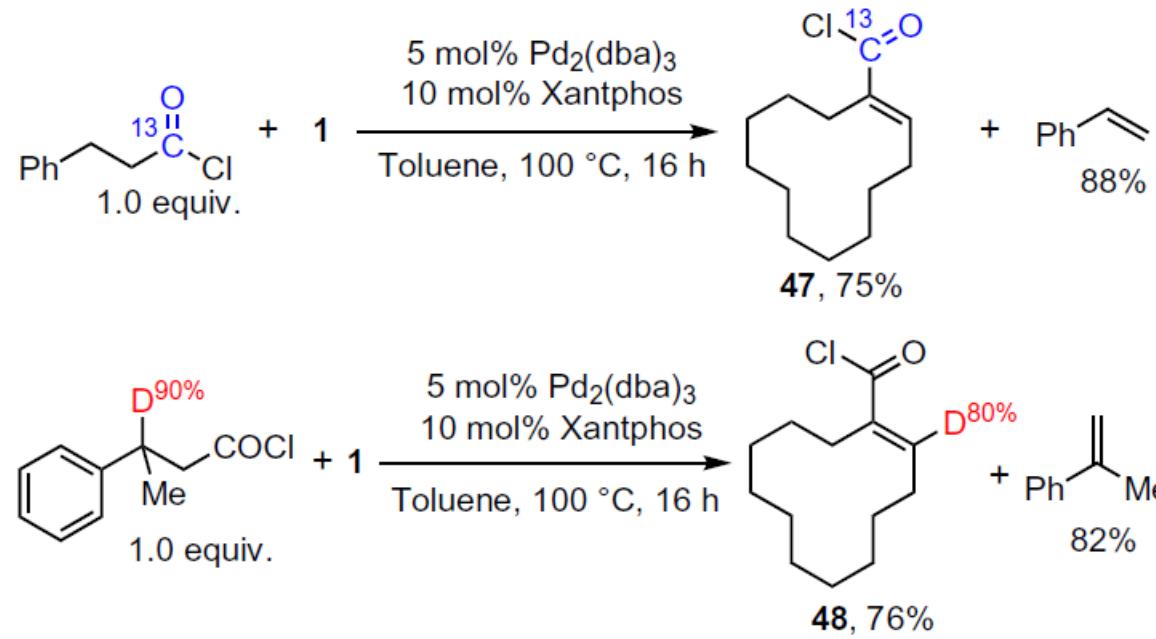


Table 2 | One-pot synthesis of carbonyl-containing products from unsaturated substrates.

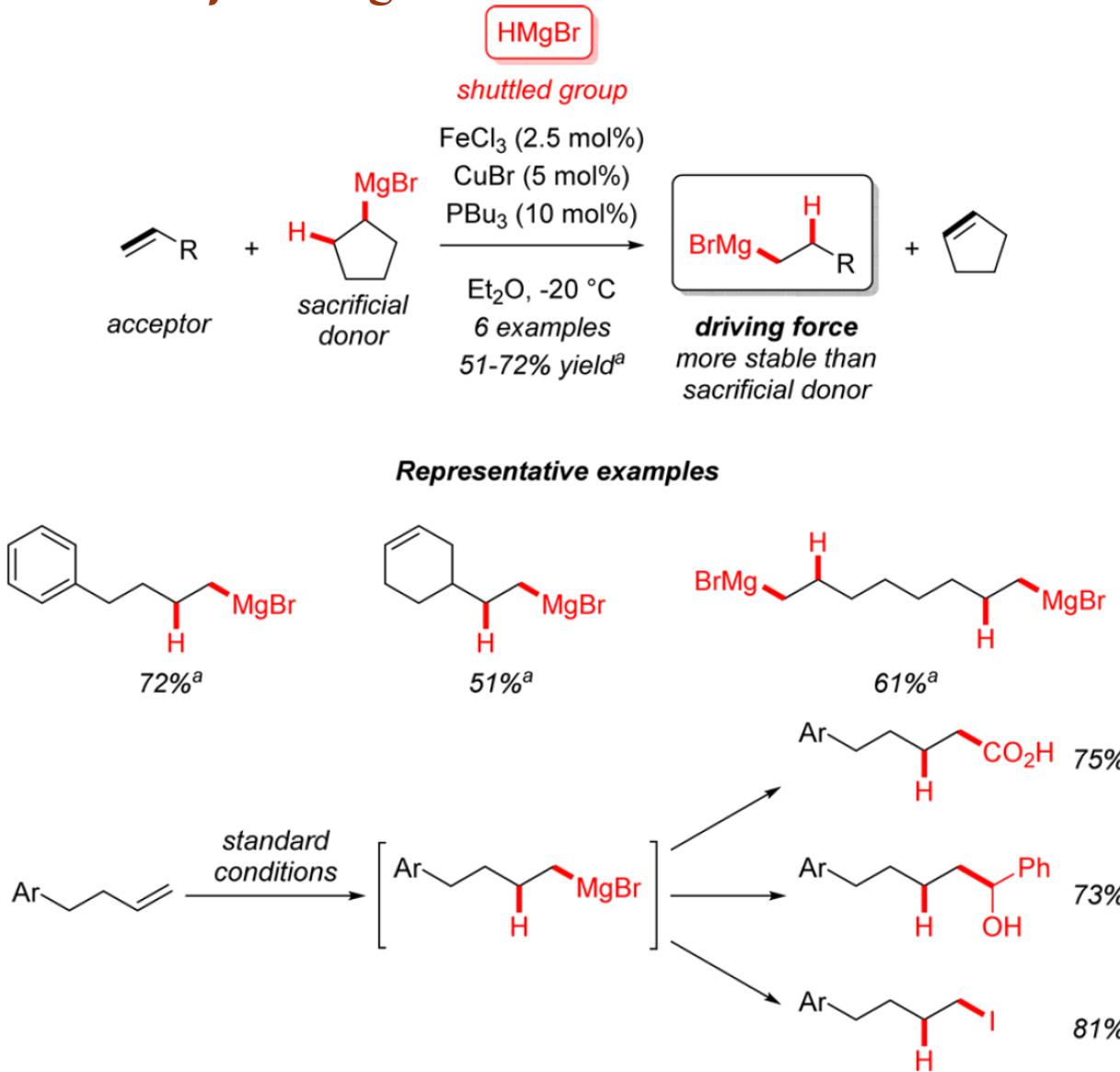
$\text{R}^1\text{---}\text{R}^2 + \text{2}$		$5 \text{ mol\% Pd}_2(\text{dba})_3$ $10 \text{ mol\% Xantphos}$ $\text{Toluene}, 100^\circ\text{C}, 16 \text{ h}$ $\xrightarrow{\text{Then conditions below}}$ One-pot	$\text{R}^1\text{---}\text{C}(=\text{O})\text{---X}$ $\text{X = OR, SR, NR}_2, \text{ hydrocarbyl}$
	32, 80%		
	33, 70%*		
	34, 66%		
	35, 80%		
	36, 67%		
	37, 65%		
	38, 58%		
	39, 55%		
	40, 71%		
	41, 82%		
	42, 46%		
One-pot Friedel-Crafts (sp^2) 1.0 equiv. SnCl_4 , CH_2Cl_2 , r.t. 1,3-Dimethoxybenzene		cat. CuI , 2.0 equiv. Et_3N , r.t.	cat. $\text{Pd}(\text{PPh}_3)_2\text{Cl}_2$, 1.2 equiv. $\text{K}_3\text{PO}_4 \cdot \text{H}_2\text{O}$, 80°C PhB(OH)_2
		$\equiv \text{Ph}$	
			cat. Fe(acac)_3 , -78°C to r.t. PhCH_2MgCl

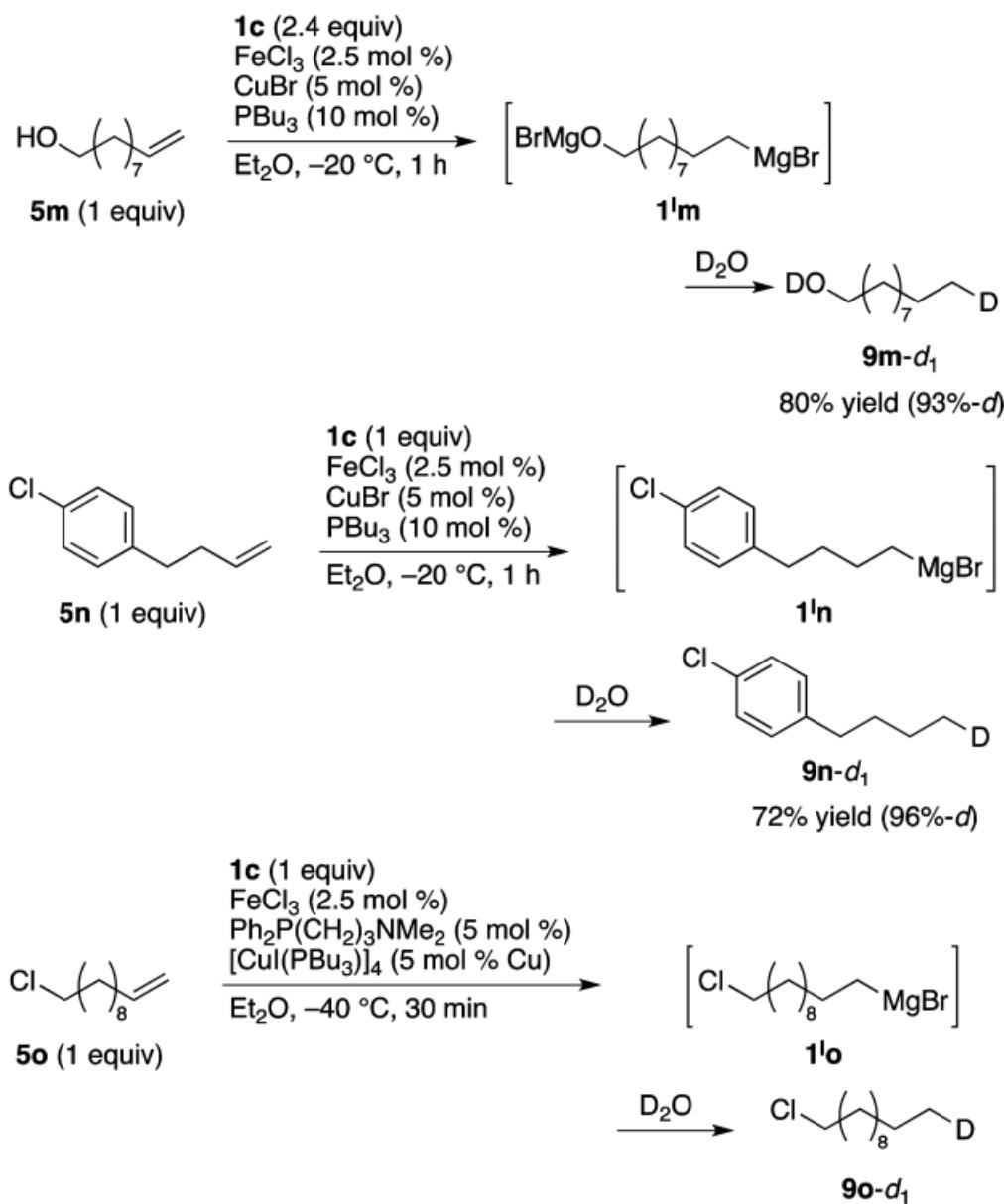
*Base (Et_3N) was added. cat., catalytic; r.t., room temperature.

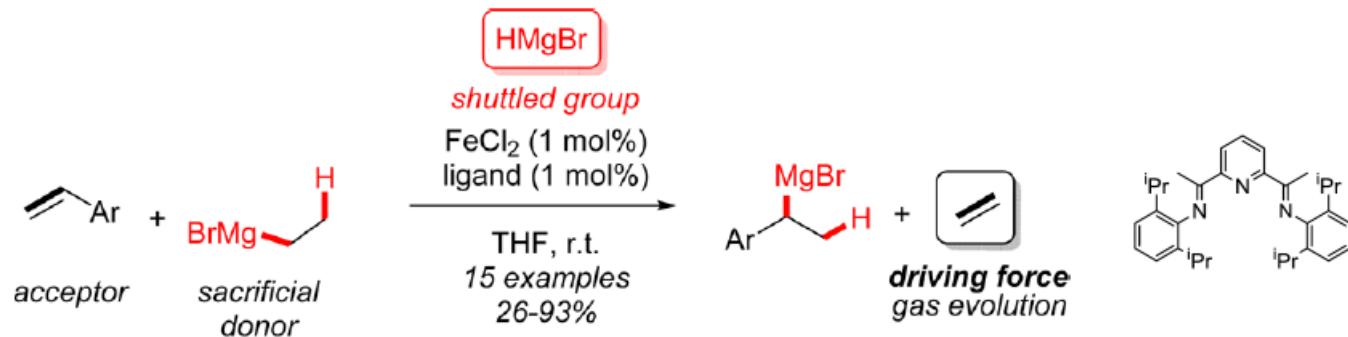




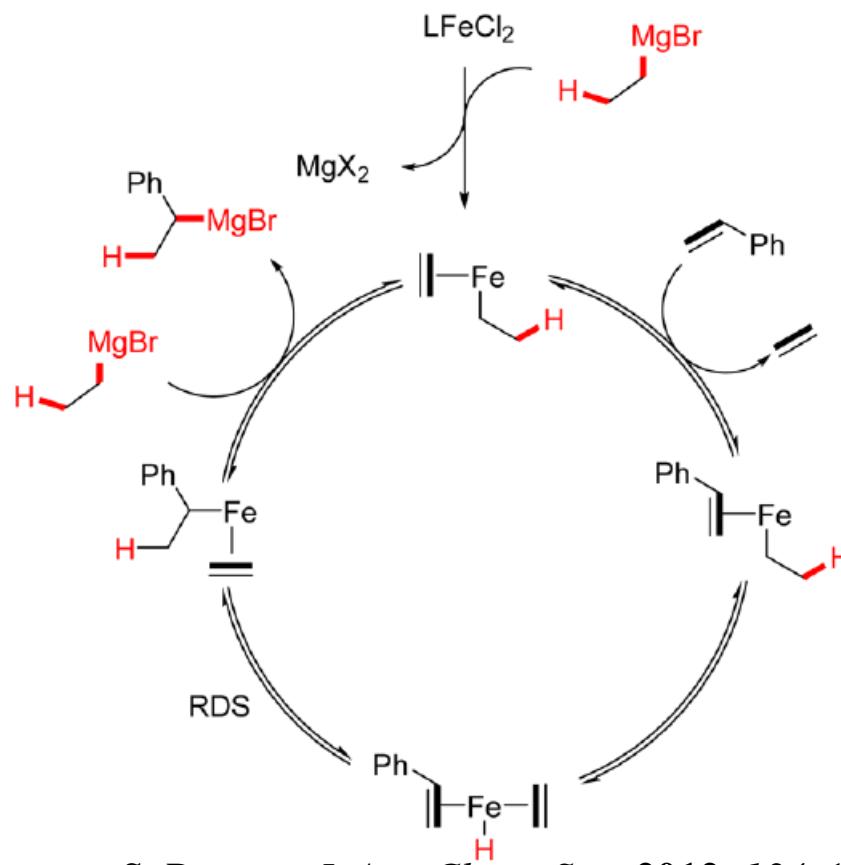
5. Catalytic Transfer Hydromagnesiation:



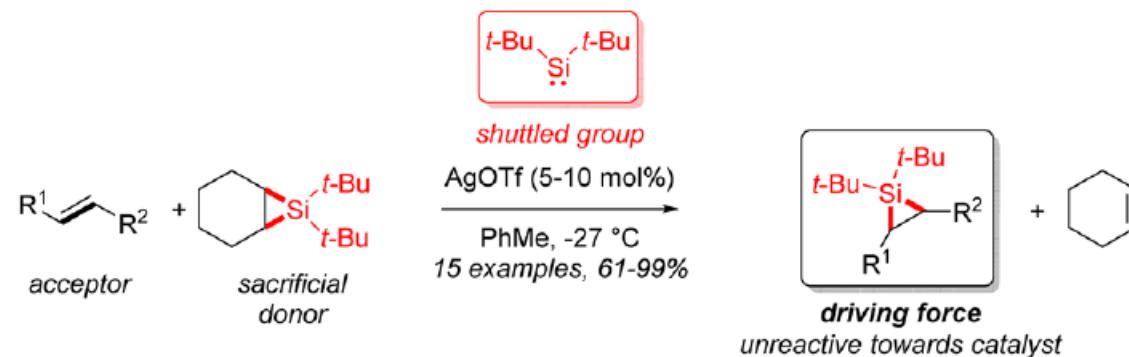




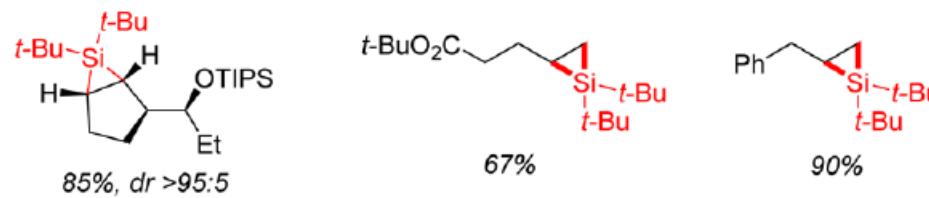
Reaction mechanism



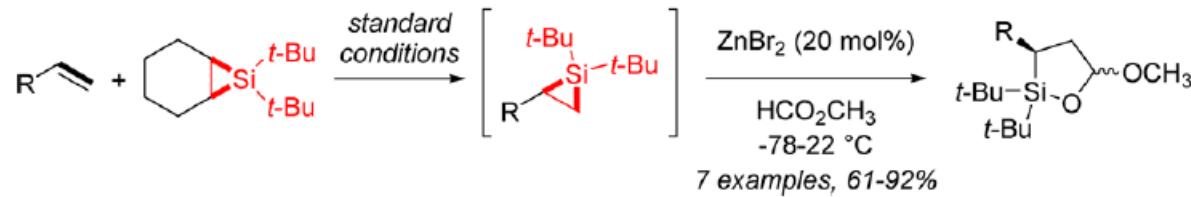
5. Catalytic Transfer Silacyclopropanation:



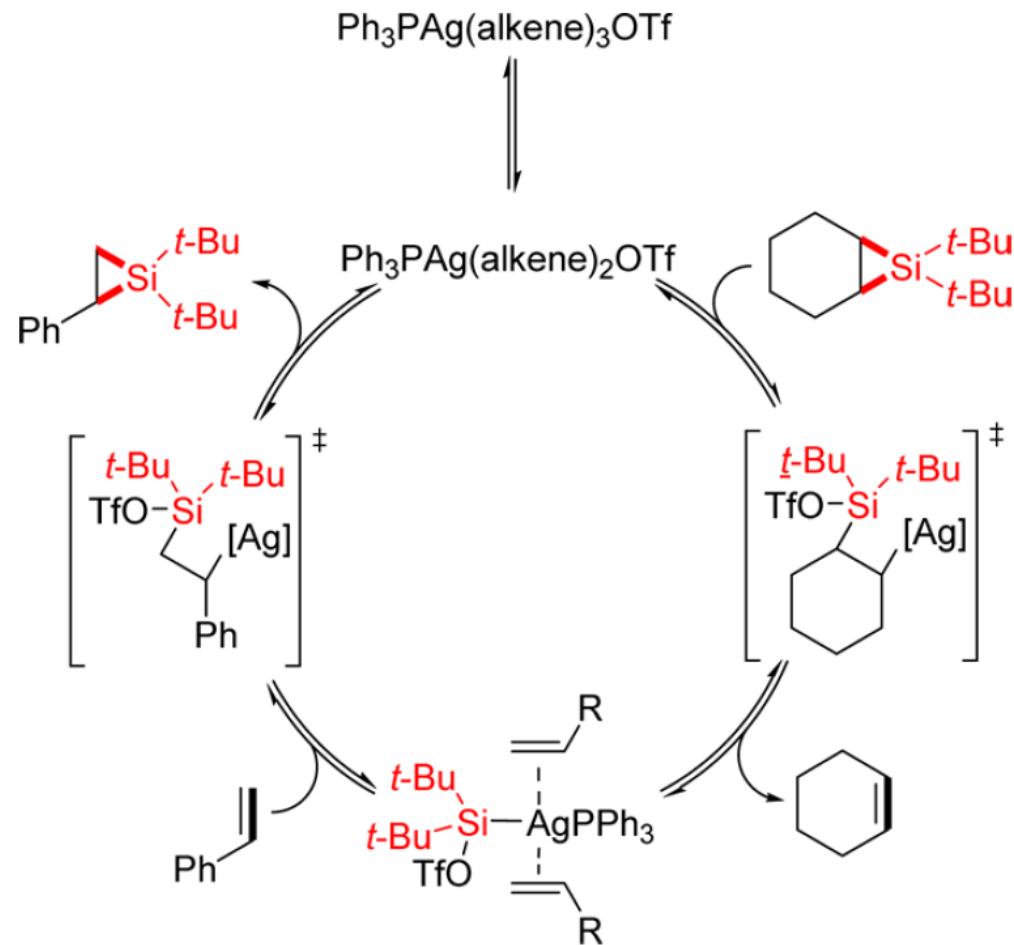
Representative examples



Application to one-flask silylene transfer/methyl formate insertion



Woerpel, K. A., et al. *J. Am. Chem. Soc.* **2004**, *126*, 9993



**Thanks for Your
Attention !**